

C O R R E S P O N D E N C E

The Use of Nearly Neat $^{16}\text{O}^{18}\text{O}$ in Spectroscopic Studies of Oxyhemerythrin and Oxyhemoglobin¹

In their interesting article on hemerythrin,² Klotz and Kurtz mention the use of nearly neat $^{16}\text{O}^{18}\text{O}$ in a resonance Raman spectroscopic study of the bonding of O_2 in oxyhemerythrin, citing unpublished work in a dissertation.³ Your readers may be interested to learn that a prior study of the O_2 bonding in oxyhemoglobin was carried out with the same $^{16}\text{O}^{18}\text{O}$ sample, and led to a similar conclusion, i.e., that the O_2 was bonded in such a way that the two oxygen atoms were not equivalent.⁴ To the best of our knowledge, this work constituted the first preparation of neat $^{16}\text{O}^{18}\text{O}$ and its first use in spectroscopy. The work represents a collaboration between the group at Northwestern and my group at Argonne National Laboratory, where the $^{16}\text{O}^{18}\text{O}$ was prepared by a rather laborious procedure that entailed the intermediate synthesis of ^{18}O -enriched hypofluorous acid.⁵

We have subsequently developed a more convenient synthesis of nearly neat $^{16}\text{O}^{18}\text{O}$ that makes use of the re-

cently isolated fluoroxysulfate ion,^{6,7} SO_4F^- , and we anticipate that this uniquely labeled oxygen molecule will now find application in a variety of spectroscopic measurements.

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(1) Based on work performed under the auspices of the Office of Energy Research, Division of Chemical Sciences, U.S. Department of Energy, under Contract W-31-109-Eng-38.

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